

Al-Ir-Ni (Aluminum-Iridium-Nickel)

V. Raghavan

Recently, [2007Yam] determined an isothermal section for this system at 1300 °C in the composition range up to 50 at.% Al.

Binary Systems

The Al-Ir phase diagram [Massalski2] depicts the following intermediate phases: Ir₂Al₉ (*D8_d*, Co₂Al₉-type monoclinic), Ir₄Al₁₃ (monoclinic), IrAl₃ (*D0₁₈*, Na₃As-type hexagonal), Ir₂Al₅ (or IrAl_{2.7}, cubic), and IrAl (*B2*, CsCl-type cubic). The Al-Ni phase diagram [Massalski2] shows five intermediate phases: NiAl₃ (*D0₁₁*, Fe₃C-type orthorhombic), Ni₂Al₃ (*D5₁₃*-type hexagonal), NiAl (*B2*, CsCl-type cubic), Ni₅Al₃ (Ga₃Pt₅-type orthorhombic), and Ni₃Al (*L1₂*, AuCu₃-type cubic; denoted γ'). Ir and Ni form a continuous face-centered cubic (fcc) solid solution.

Ternary Isothermal Section

[2007Yam] arc-melted under Ar atm 18 ternary alloys (including 4 alloys investigated by [2003Yam]), with Al

content up to 50 at.%. The alloys were annealed at 1300 °C. The phase equilibria were studied with x-ray diffraction, scanning and transmission electron microscopy, and electron probe microanalysis. The isothermal section at 1300 °C constructed by [2007Yam] is shown in Fig. 1. NiAl and IrAl form a continuous *B2* solid solution. The two-phase equilibrium between fcc and *B2* phases dominates the section. The *L1₂* (Ni₃Al) phase dissolves up to 3.5 at.% Ir. A narrow three-phase region of (*L1₂* + fcc + *B2*) is present.

References

2003Yam: Y. Yamabe-Mitarai and H. Aoki, Phase Constitution and Creep Properties of Ir-Ni-Al Alloys, *Mater. Sci. Eng. A*, 2003, **A362**, p 152-159

2007Yam: Y. Yamabe-Mitarai, T. Aoyagi, K. Nishida, H. Aoki, T. Abe, and H. Murakami, Phase Equilibria between the *B2*, *L1₂*, and fcc phases in the Ir-Ni-Al System, *Intermetallics*, 2007, **15**, p 479-488

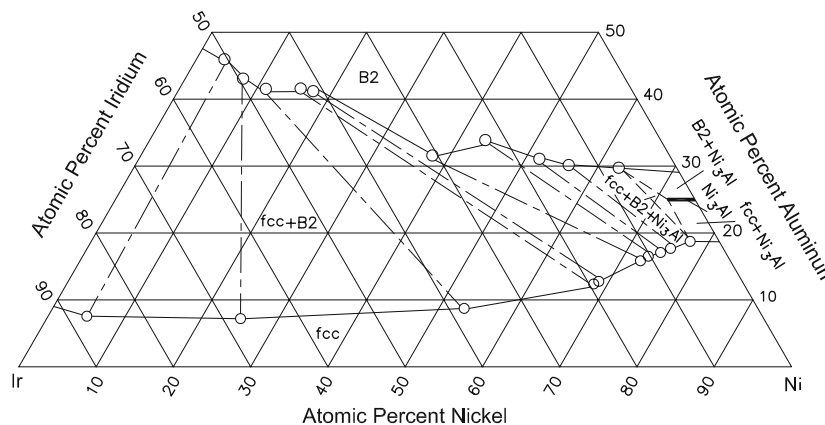


Fig. 1 Al-Ir-Ni isothermal section at 1300 °C [2007Yam]